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(54) **Wash cycle or rinse cycle fabric conditioning compositions.**

(57) Fabric conditioning aqueous liquid emulsions are provided which impart softness and antistatic properties to laundered fabrics as a wash-cycle additive for through-the-wash use or alternatively as a rinse cycle additive. The dispersed phase of the liquid emulsions comprise complexes of specified tertiary amines and multi-functional carboxylic acids in combination with unreacted amine and a minor amount of one or more specified alkyl cellulose ethers to enhance high temperature emulsion stability.

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BACKGROUND OF THE INVENTION

This invention relates to novel aqueous liquid emulsions useful as through-the-wash or rinse cycle-additive fabric conditioning compositions and their method of manufacture, such fabric conditioning compositions providing softening and antistatic benefits to laundered fabrics without adversely affecting cleaning.

A large number of compositions have been disclosed which impart softening and antistatic properties to laundered fabrics. Generally, these contain cationic compounds, especially quaternary ammonium salts. Such compositions are widely marketed for home use in the form of emulsions which must be added to the washing machine during the rinse cycle. If the emulsions are added during the wash cycle, the cationic fabric conditioners may interact with anionic surfactants present in the washing composition so as to render a portion of each of such cationic compound and anionic surfactant unavailable for either cleaning or fabric conditioning.

Another means of providing fabric conditioning which has attained some commercial success is to add the conditioning agent while the clothes are being machine dried.

While fabric conditioning during either the rinse and/or drying cycles can be effective, both methods of conditioning are more inconvenient than a through-the-wash method where the conditioning agent is added with the detergent composition at the initiation of the wash cycle.

Compositions are known which can be added to a washing machine at the start of the wash cycle and effectively provide fabrics with a detergency treatment during the wash cycle and a fabric conditioning treatment during either the rinsing operation or subsequently when the fabrics are heated in a machine dryer. Compositions of this type are known in the art as through-the-wash fabric conditioners. An important advantage of such compositions is that they obviate the need for adding a separate fabric conditioning product in the rinse cycle or in a machine dryer.

Through-the-wash type conditioning agents are well known in the art. European Patent Application No. 0,123,400, published October 31, 1984 discloses fabric conditioning agents comprising salts of specified tertiary amines and carboxylic acids which are utilized in the form of nodules which pass virtually unchanged through the wash and rinse and condition the fabric when heated in a dryer. The nodules are meant to be added to the laundry wash liquor at the beginning of the wash cycle along with a conventional detergent. European Patent Publication No. 0,133,804 published March 6, 1985 discloses detergent compositions containing clay fabric softeners and particles of a complex of a long chain amine and a fatty acid. U.S. Patent No. 4,514,444 to Ives discloses a fabric cleaning/conditioning composition comprising carboxylic acid salts of a tertiary amine in combination with polyethylene glycol. U.S. Patent No. 4,375,416 to Crisp et al. discloses a textile softening detergent composition comprising a specified class of tertiary amines with a smectite-type clay in a detergent composition such that softening benefits are provided without impairing cleaning performance.

Other recent prior art relating to the field of the invention includes U.S. Patent No. 4,237,155 to Kardouche which discloses a dryer-added fabric conditioning agent comprised of a carboxylic acid salt of a tertiary amine. British Patent 1,514,276 discloses the use of tertiary amine compositions as wash-cycle fabric softeners.

Complexes of specified tertiary amines and carboxylic acids have been previously disclosed as through-the-wash fabric conditioners in U.S. Patent No. 4,828,722 to Steltenkamp. In U.S. Patent 4,869,836 to Harmalker there is described the combination of unreacted tertiary amine with a complex of reacted tertiary amine and multi-functional carboxylic acid. The resulting fabric conditioners are used advantageously in the form of aqueous liquid emulsions. While these emulsions are generally effective for fabric conditioning, they nevertheless are often destabilized or suffer poor performance when subjected to extreme conditions, such as elevated temperatures, i.e. above 110° F. Consequently, there remains a need for providing an effective fabric-conditioning liquid emulsion capable of being used as either a wash-cycle or rinse-cycle additive, and which is highly stable at elevated temperatures.

SUMMARY OF THE INVENTION

The present invention provides a wash cycle or rinse cycle-additive aqueous liquid emulsion for providing softness and antistatic properties

to fabrics treated therewith in a laundry bath without adversely affecting fabric cleaning comprising

(A) from about 1 to 30%, by weight of a particulate fabric conditioning composition having a median particle diameter greater than about 10 microns comprising:

(a) a fabric conditioning amount of a multi-functional carboxylic acid complex of a tertiary amine

formed from the reaction of (i) a tertiary amine having the general formula:



wherein R<sub>1</sub> is methyl or ethyl, and R<sub>2</sub> and R<sub>3</sub> are each independently an aliphatic group having from 12 to 22 carbon atoms, and (ii) a multi-functional carboxylic acid selected from the group consisting of citric acid, and di and tricarboxylic acids having from 21 to 54 carbon atoms;

(b) unreacted tertiary amine having the general formula defined above, said unreacted amine being present in an amount of at least 0.2 times the stoichiometric amount of tertiary amine required to form the multi-functional carboxylic acid complex of (a); and

(c) an alkyl cellulose ether selected from the group consisting of methylcellulose, hydroxypropyl-methyl cellulose and derivatives of hydroxyethyl cellulose wherein the terminal hydrogen of the hydroxyether group is replaced by an alkyl chain having from 10 to 24 carbon atoms, and mixtures thereof, the dispersed phase of the liquid emulsion being essentially comprised of said particles of fabric conditioning composition.

(B) from about 0.1 to 10%, by weight of one or more emulsifying agents; and

(C) the balance water.

The most effective fabric conditioning compositions of the invention contain an amount of unreacted tertiary amine varying from about 0.3 to 6 times the stoichiometric amount of reacted amine in the complex, preferably from about 0.6 to 6 times the said stoichiometric amount. Where, for example, the multi-functional carboxylic acid selected to form the complex is citric acid, the stoichiometric amount of reacted amine in the complex is 3 moles of amine per mole of citric acid.

The term "complex" as used throughout the specification and claims refers to the reaction product of the above described tertiary amine and carboxylic acid, and characterizes such reaction product in terms of the primary constituent thereof which is a complex rather than a salt of the acid and amine. The basis of such characterization is explained hereinafter in the specification. Although the applicant does not wish to be limited by any theory regarding the nature of such reaction product, it is believed to be an equilibrium mixture comprised of the acid-amine complex (about 80%, by weight) and the acid-amine salt (about 20%, by weight). Accordingly, as used herein, the term "complex" includes both the acid-amine complex formed by the reaction of the tertiary amine and carboxylic acid as well as the relatively minor amount of salt in equilibrium therewith.

In accordance with a preferred embodiment of the invention, the aqueous liquid emulsion contains adjuvants such as perfumes, colorants, brighteners, foam stabilizers and the like and, optionally further includes an antistatic composition distinct from the defined fabric conditioning composition to enhance the anti-static properties of the liquid emulsion. A preferred additional anti-static composition for this purpose is tallow neodecanamide.

In accordance with the process aspect of the invention, softness and anti-static properties are imparted to fabrics by contacting such fabrics in a laundry bath or rinse liquor with an aqueous liquid emulsion containing an effective amount of a particulate fabric conditioning composition having a median particle diameter greater than about 10 microns, which fabric conditioning composition comprises:

(a) a fabric conditioning amount of a multi-functional carboxylic acid complex of a tertiary amine formed from the reaction of (i) a tertiary amine having the general formula:



wherein R<sub>1</sub> is methyl or ethyl, and R<sub>2</sub> and R<sub>3</sub> are each independently an aliphatic group having from 12 to 22 carbon atoms, and (ii) a multi-functional carboxylic acid selected from the group consisting of citric acid, and di and tricarboxylic acids having from 21 to 54 carbon atoms;

(b) unreacted tertiary amine having the general formula defined above, said unreacted amine being present in an amount of at least 0.2 times the stoichiometric amount of tertiary amine required to form

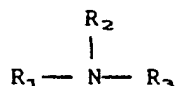
the multi-functional carboxylic acid complex of (a); and

(c) an alkyl cellulose ether selected from the group consisting of methylcellulose, hydroxypropylmethyl cellulose and derivatives of hydroxyethyl cellulose wherein the terminal hydrogen of the hydroxyether group is replaced by an alkyl chain having from 10 to 24 carbon atoms, and mixtures thereof.

Although the applicants do not wish to be bound by any theory of operation, it is believed that the alkyl cellulose ether incorporated into the liquid emulsion of the invention forms at least a partial coating upon the particles of fabric conditioning composition which are formed upon emulsification. This coating appears to substantially prevent particle break-down over a wide range of temperatures as well as the undesired subsequent coalescence of smaller particles of acid-amine complex into larger aggregate particles, a problem characteristic of liquid emulsions known in the art containing particulate complexes of carboxylic acid and amine which are typically susceptible to high temperature breakdown and concomitant product separation. It has further been discovered that for particulate compositions having a median particle diameter above about 10 microns, and more preferably at median particle sizes from about 25 to 50 microns, the fabric conditioning, physical stability and flow characteristics of the resulting liquid emulsion are at an optimum. Accordingly, for the fabric conditioning compositions of the invention the desired range of particle size is able to be maintained over a broad range of temperature extending from ambient to above the melting point of the carboxylic acid - amine complex, typically about 110° F.

#### DETAILED DESCRIPTION OF THE INVENTION

The fabric conditioning compositions of the invention are comprised of three essential components. The first and second components are, respectively, a complex of tertiary amine with a multi-functional carboxylic acid as herein defined and unreacted tertiary amine. The suitable tertiary amines are represented by the general formula



wherein R<sub>1</sub> is methyl or ethyl, and R<sub>2</sub> and R<sub>3</sub> are each independently an aliphatic group having from 12 to 22 carbon atoms. Examples of preferred amines include methyl distearyl amine, ethyl distearyl amine, methyl di(hydrogenated tallow) amine, ethyl di(hydrogenated tallow) amine, methyl diolelylamine, methyl

dicoconut amine, methyl dilaurylamine, and methyl dipalm oil amine. The multi-functional carboxylic acid utilized in the present invention is selected from among citric acid and di and tri carboxylic acids having 21 to 54 carbon atoms. Most preferred for use herein is citric acid. Among the other preferred acids are a dicarboxylic acid having 21 carbon atoms e.g. 5 (or 6)-carboxy-4 hexyl-2-cyclohexene-1-octanoic acid (sold commercially under the tradename Westvaco Diacid 1550 by Westvaco Corporation); dimerized oleic acid (sold commercially under the tradename Dimer Acid by Emery Industries); and a C<sub>54</sub> trimer of oleic acid.

The amine-multifunctional carboxylic acid complexes of the invention are generally prepared by forming a mixture of amine and multicarboxylic acid, preferably in a molar ratio of amine to carboxylic acid above that required for the stoichiometric reaction so as to provide the desired amount of unreacted amine in the reaction product, and heating such mixture to a temperature sufficient to form a melt.

For the example of a tertiary amine having a melting point below that of the carboxylic acid, preparation is conveniently effected by first heating the amine to its melting point (generally about 35 to 45° C) and then adding thereto the multicarboxylic acid, such as for example citric acid, in the form of a solid. The resulting mixture is then heated to a temperature below the melting point of the carboxylic acid, but sufficient to form a molten mixture. In the case of citric acid, heating the reactant to a temperature of about 115° C for about five to ten minutes will form a molten mixture having a melting point (about 50° C) intermediate of the citric acid and the amine. The molten mixture comprises the reaction product of amine-multicarboxylic acid complex in equilibrium with a minor amount of amine-carboxylic acid salt.

In an alternate embodiment, the amine and carboxylic acid are reacted in a stoichiometric ratio to form the complex followed by the addition of unreacted tertiary amine to the resulting reaction product. This embodiment is particularly advantageous where it is desired to employ an unreacted amine in the fabric condition composition which is different from the reacted amine. This may be desirable in some instances for purposes of economy.

The determination of the nature of the reaction product can be illustrated in terms of the reaction between methyl di(hydrogenated tallow) amine and dimerized oleic acid which were mixed and heated following the general procedure described above except that in this instance the amine having a melting point above the carboxylic acid is added in solid form to dimerized oleic acid which is liquid at ambient temperature. The resulting reaction product was identified as a weak hydrogen bonded complex (80 wt.%) in equilibrium with the corresponding salt (20 wt.%). Identification was based on measurements involving melting points and spectroscopic techniques. The complex melted at 28 to 31 °C which is intermediate between the melting point of the amine (34 to 38 °C) and the carboxylic acid (4 to 5 °C). This indicates the formation of a complex rather than an amine salt, the latter being characterized by a sharp melting point higher than the corresponding amine.

The infra red spectrum of the complex shows the presence of two moderate carbonyl bands at wavelengths of 1709  $\text{cm}^{-1}$  and 1550  $\text{cm}^{-1}$ . The 935  $\text{cm}^{-1}$  wavelength indicative of H-bonding of the particular free carboxylic acid is absent, indicating the presence of a complex rather than salt formation. By means of ESCA (Electron Spectroscopy for Chemical Analysis) measurements, it was determined that the reaction product was about 20% amine salt and 80% of the amine-carboxylic acid complex. The chemical shift of the ionic nitrogen of the salt was different than that of the neutral nitrogen of the complex. The relative amounts of these two nitrogen signals provide the basis for determining the relative amount of amine salt versus amine complex.

An alkyl cellulose ether is the third essential component of the particulate fabric conditioning composition. Suitable alkyl cellulose ethers are selected from among methylcellulose and hydroxypropylmethyl cellulose, sold under the trademark "Methocel" by Dow Chemical Company; hydroxypropylcellulose, sold under the trademark "Klucel" by Hercules Chemical Company; and derivatives of hydroxyethyl cellulose (HEC) wherein the terminal hydrogen of the hydroxyether group is replaced by an alkyl chain having from 10 to 24 carbon atoms, such HEC derivatives being sold under the trademark "Natrosol Plus" by Hercules Chemical Company, and are extensively described in U.S. Patent 4,228,277 issued October 14, 1980. The amount of alkyl cellulose ether present in the particulate fabric conditioning composition is generally from about 0.5 to 10%, by weight, of the particulate composition, and preferably, from about 1 to 5%, by weight, in order to provide the requisite high temperature stability to the particles of fabric conditioning composition.

The aqueous liquid emulsion of the invention may be advantageously added to the laundry bath or to the rinse liquor independent of any laundry detergent composition or may conveniently be added to the laundry bath during the wash cycle in conjunction with a liquid or granular detergent composition.

The method of preparation of the aqueous emulsion is predicated upon forming an emulsion or suspension which is stable over a practical range of temperatures, and particularly at high temperature, namely, it does not undergo phase separation at temperatures up to about 120 °F, and, in addition, the particles of fabric conditioning composition which comprise the dispersed phase of the emulsion must be of the requisite size to deposit on washed fabrics during the wash cycle. It has been discovered that when present in an emulsion at particle sizes having a median diameter above about 10 microns, preferably from about 25 to 100 microns, and most preferably from about 25 to 50 microns, the composition of the invention is capable of providing effective softening and anti-stat properties to washed fabrics whereas at particles sizes below such value fabric conditioning is often adversely affected. Although the applicant does not wish to be bound by any theory, it is believed that particle sizes of above about 10 microns are required in order to effect deposition of said particles on fabrics in the wash water, such particle size limitation not being critical for rinse cycle softening.

To insure high temperature stability and the avoidance of phase separation, the HLB (hydrophilic-lipophilic balance) value of the emulsion is preferably regulated to within a predetermined range required for stability by the addition of suitable emulsifying agents. The required range of HLB is readily determined by trial and error for each particular combination of tertiary amine and carboxylic acid utilized in the composition of the invention. For the particular instance where citric acid and methyl di(hydrogenated tallow) methyl amine are used to form the fabric conditioning composition, the HLB of the emulsion must be from about 11.5 to 12.5 to achieve the desired high temperature stability.

The liquid emulsion compositions preferably contain from about 5 to about 30%, and most preferably from about 7 to about 20% of the fabric conditioning composition of the invention, based on the total weight of the emulsion composition. At such concentrations, an effective amount of fabric conditioning composition is provided to the wash fabric when dispensing an amount of the liquid emulsion to the wash or rinse cycle of an automatic washing machine comparable to the amounts added by users of commercial liquid fabric conditioners.

Nonionic surfactants are among the preferred emulsifying agents for preparing an emulsion in accordance with the invention having the desired stability, viscosity and particle size of fabric conditioning

composition in the dispersed phase. Among the useful emulsifying agents are Neodol 25-3 (an ethoxylated alcohol sold by Shell Chemical Company comprising a fatty alcohol averaging about 12 to 15 carbon atoms with about 3 moles of ethylene oxide per mole of alcohol); Neodol 25-12; and Neodol 45-13. Neodol 25-3 and 45-13 are particularly preferred for this purpose.

5 The preparation of the emulsion is conveniently effected in three stages: in the first stage the fabric conditioning composition, preferably at a temperature above its melting point, is added to an aqueous liquid, preferably water, along with a first portion of an emulsifying agent, such first portion being an amount selected to form upon mixing with the fabric conditioning composition particles of emulsified fabric conditioning composition having a median diameter above about 10 microns. The order of addition of the  
10 fabric conditioning composition and the first portion of emulsifying agent is not critical. It is preferred that the aqueous liquid be preheated to a temperature corresponding to at least the melting point of the fabric conditioning composition if the latter is introduced as a liquid. This is to insure that the emulsified particles formed in the first stage are in liquid form. In an alternate embodiment, the fabric conditioning composition is introduced into the aqueous liquid as a solid, following which the liquid is heated to a temperature  
15 sufficiently above the melting point of the conditioning composition such that upon mixing the conditioning composition with the first portion of emulsifying agent, there is provided an emulsion containing as the dispersed phase liquid particles of fabric conditioning composition having the desired particle size.

In the second stage the resulting emulsion is cooled to a temperature sufficiently below the melting point of the fabric conditioning composition so as to at least partially solidify the emulsified particles and  
20 form a suspension of solid particles in the aqueous liquid.

In the third stage, a second portion of one or more emulsifying agents is added to the emulsion or suspension formed in the second stage so as to adjust the HLB value to that required for high temperature stability. This HLB value can be conveniently determined by a simple trial and error technique. As defined herein, the characterization of "high temperature stability" for a liquid emulsion in accordance with the  
25 invention refers to its being able to be maintained at 120° F for at least 24 hours without the occurrence of phase separation. After the formation of the emulsion in the third stage, electrolytes such as calcium chloride dihydrate, or sodium chloride may be added as viscosity modifiers, if needed, as well as defoaming materials to enhance proper mixing of the components by inhibiting phase separation resulting from foam agitation. Other optional components include colorants and perfume which are advantageously  
30 added sequentially under agitation.

The emulsified particles in the dispersed phase of the emulsion are not all of uniform size and comprise a broad distribution of particle sizes, but it is required that the median diameter of such particles be above 10 microns. A preferred particle size is that having a mean diameter of from about 25 to 50 microns. Measurement of the emulsified particles is most conveniently carried out at the end of the third stage when  
35 the final emulsion is formed rather than at the end of the first stage where the relatively strong association of the emulsified particles may make the particle size measurement somewhat less accurate.

The aforementioned three-stage method of preparation is predicated upon utilizing a fabric conditioning composition having a melting point above ambient temperature such that in the first stage of preparation only a limited amount of emulsifying agent is added to provide the desired size of emulsified particles as a  
40 dispersed liquid phase. Thereafter upon cooling, the dispersed particles solidify, allowing additional amounts of emulsifying agent to be added to the emulsion without causing any diminution in particle size. Thus, the HLB of the emulsion can be independently adjusted to the desired range without affecting the size of the particles in the dispersed phase.

#### 45 EXAMPLE 1

A liquid emulsion in accordance with the invention was prepared as follows:

To 0.3 grams of Methocel A4C<sup>R</sup>, a methylcellulose marketed by Dow Chemical Company, there was added 5 grams of deionized water at 70° C and mixed to form a paste. This paste was added to 79.46  
50 grams of water at 25° C and thoroughly mixed to obtain a uniform dispersion, which was then heated to 43° C.

Amine citrate complex was prepared by adding 0.65 grams of anhydrous citric acid to 9.35 grams of melted methyl di(hydrogenated tallow) amine (sold commercially as Armeen M2HT by Akzo Chemicals Incorporated) corresponding to a molar ratio of tertiary amine to citric acid of 5.2 to 1 and the mixture  
55 heated till all the citric acid is melted and dissolved in the amine (approximate temperature: 115° C). The solution was then cooled to room temperature where it solidified.

The amine citrate complex plus the unreacted amine was thereafter melted by heating to about 70° C and added to the methylcellulose dispersion described above at 43° C under gentle agitation to form a

smooth, white emulsion having a cream-like consistency. This emulsion was cooled to approximately 40° C and 0.2 grams of Neodol 45-13 (a tradename for a Shell Chemical Company detergent which is a condensation product of a mixture of fatty alcohols averaging about 14 to 15 carbon atoms with about 13 moles of ethylene oxide per mole of alcohol) dissolved in 0.8 grams water was added to it while mixing. The emulsion was gradually cooled to 25° C to let the particles solidify.

The particle size of the dispersed phase as determined by a HIAC/ROKO Particle Size Analyzer (Model PA 720) marketed by Pacific Scientific Company was about 35 microns mean diameter. Such Analyzer uses established light blocking principles for measuring the particle size mean diameter whereby the measured particles interrupt a continuous light beam when passing through a sensing zone which, in turn, causes a reduction in the amount of light reaching a photo detector. This technique is particularly advantageous for measuring particle sizes larger than the normal range of particles conventionally measured by light scattering techniques.

To this solution there was added, while mixing, 0.6 grams of Neodol 45-13 dissolved in 2.4 grams water followed by 0.35 grams of Neodol 25-3. The HLB of the emulsion following the addition of these emulsifying agents was about 12.

A commercial defoaming material (0.02 grams) was then added to the resulting emulsion followed by the addition of adjuvants such as perfume (0.5 grams) dye (.015 grams) and preservative (0.1 gram of Ucaricide marketed by Union Carbide Corporation), all while mixing. A minor amount of silicone (0.25 grams of Dow Corning 193) was then added to modify the viscosity.

The resulting emulsion was highly stable over a temperature range from ambient to at least 110° F, and, in particular, did not manifest any particle break down and phase separation when aged at high temperature, namely a minimum of 24 hours at a temperature of 110° F or above.

#### EXAMPLE 2

A liquid emulsion of the invention was prepared utilizing as the alkyl cellulose ether a derivative of hydroxyethyl cellulose (HEC) in which the hydroxyl hydrogen of the ethyl hydroxyl group on the 5th carbon of the ring is replaced by a fatty alkyl chain having from 10 to 24 carbon atoms. Such HEC derivative polymers are sold under the trademark "Natrosol Plus" by Hercules Chemical Company.

An amine-citrate complex was prepared by adding 0.654gms of anhydrous citric acid crystals to 9.34gms of molten methyl-di (hydrogenated tallow) tertiary amine (sold commercially as Armeen M2HT by Akzo Chemicals, Inc.) and the mixture was maintained at a temperature of about 115° C until the citric acid complexed with the amine. The resulting binary mixture with excess-free amine was cooled to room temperature under ambient conditions to obtain a white solid cake.

Deionized water (89.33gms) was heated to about 48° C using a bench top hot plate. Amine-citrate complex (9.994gms) prepared according to the aforementioned procedure was melted in a microwave oven and added slowly to the above hot water under gentle agitation using a paddle mixer. The mixing was continued for about 10-15 minutes and the resulting milky emulsion was cooled to 40° C and maintained at this temperature.

To this amine-citrate emulsion there was then slowly added 0.06 grams of the above described HEC derivative polymer while maintaining gentle agitation. The emulsion was mixed for an additional 10 minutes and then allowed to cool to ambient temperature (approximately 25° C). To the above cooled emulsion 0.2 grams of liquid Neodol 25-13 (marketed by Shell Chemical Company) was added followed by the addition of color and perfume to complete the prototype preparation. The above method of preparation resulted in a smooth and milky emulsion having cream like consistency.

The particle size of the emulsion was measured using Olympus BM-2 microscope and the average particle size was found to be in the range of 15-25 microns. The emulsion did not separate or manifest any particle disintegration when heated to a temperature of about 110° F. This was confirmed by a hot-stage video-microscopy. The emulsion was stable over a range of temperature from ambient to 110° F. The emulsion was also stable under heat-cool (110° F to ambient) cycling conditions.

#### EXAMPLE 3

To demonstrate the improved stability of a fabric conditioning liquid emulsion in accordance with the invention, a comparison was made with the emulsion described in U.S. Patent No. 4,869,836 to Harmaker, in Example 5, the disclosure of which is incorporated herein by reference. The fabric conditioning compositions described in the Harmaker patent are similar to that described herein except for the absence of an alkyl cellulose ether.

The particle size of an emulsion prepared as set forth in Example 1 above was measured at room temperature with the HIAC/ROKO particle size analyzer. The particle size was about 35 microns mean diameter. No change in particle size was noted after about 24 hours. A second sample of the same emulsion was maintained at a temperature of about 110° F for 24 hours and its particle size then measured.

5 The resulting particle size was unchanged, about 35 microns mean diameter.

By way of comparison, a liquid emulsion was prepared in accordance with Example 5 of U.S. 4,869,836 and its measured particle size at room temperature was about 35 microns mean diameter. No change in particle size was noted after about 24 hours. A second sample of this same emulsion was maintained at a temperature of about 110° F for 24 hours and the resulting measured particle size was below 10 microns mean diameter, indicating instability of the emulsion at elevated temperatures.

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#### EXAMPLE 4

A commercial granular detergent composition designated herein as Control "A" was used in this example and had the following composition:

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#### Control A

	<u>Component</u>	<u>Weight Percent</u>
20	Linear alkylbenzene sulfonate	4
	Sodium fatty alcohol sulfate	9
25	Sodium ethoxy alcohol sulfate	3
	Polyethoxylated alcohol	0.7
30	Pentasodium tripolyphosphate	31
	Sodium pyrophosphate	7
	Sodium carbonate	9
35	Sodium sulfate	16
	Sodium silicate	5
40	Moisture and adjuvants	Balance

The following washing procedure was used to evaluate the efficacy of a composition in accordance with the invention (the composition of Example 1) and a comparative composition as described in the aforementioned U.S. Patent 4,869,836, Example 5 thereof. Each of the aforementioned liquid compositions in an amount of 90 grams was added along with 86 grams of Control A to a U.S. top-loading washing machine. A 6 1/2 lb ballast wash load comprised of cotton and synthetic fabrics was washed with 64 liters of water at 90° F using a fourteen minute wash cycle with rinse and spin operations followed by drying for one hour in an electric dryer. The washing and drying steps were then repeated and following the second drying operation the fabrics were evaluated for their anti-static properties by visual inspection. The terry towels in each wash load were then equilibrated to 40% humidity overnight and the following day were evaluated for softness by a six member panel. The results of the static and softness evaluation for each of the tested compositions is described in Table 1.

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TABLE 1  
PERFORMANCE OF FABRIC CONDITIONING COMPOSITIONS OF THE INVENTION

<u>Formulation</u>	<u>Softness (a)</u>	<u>Static (b)</u>
Control A	--	Very heavy
Control A + comparative composition Of U.S. Patent 4,869,836	+3	Light to none
Control A + composition of invention (Example 1)	+4	None

conditions: wash cycle, 90°F for 14 minutes; base composition: 84g of Control A

(a) Softness: difference in softness measured is based on a scale of 1 (very harsh) to 10 (very soft) relative to Control A as evaluated by a six-member panel. A difference of one unit or greater is considered significant. Control A provided a softness of from 3 to 4 when evaluated, on an absolute basis, on a scale of 1 to 10.

(b) The anti-static properties were characterized visually.

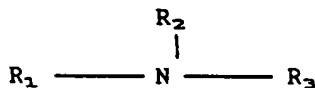
## Claims

1. A wash cycle or rinse cycle-additive aqueous liquid emulsion for providing softness and anti-static properties to fabrics treated therewith in a laundry bath without adversely affecting fabric cleaning comprising

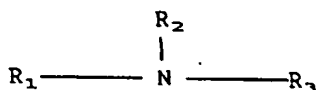
(A) from about 1 to 30%, by weight of a particulate fabric conditioning composition having a median particle diameter greater than about 10 microns comprising:

(a) a fabric conditioning amount of a multi-functional carboxylic acid complex of a tertiary amine

formed from the reaction of (i) a tertiary amine having the general formula:



- wherein R<sub>1</sub> is methyl or ethyl, and R<sub>2</sub> and R<sub>3</sub> are each independently an aliphatic group having from 12 to 22 carbon atoms, and (ii) a multi-functional carboxylic acid selected from the group consisting of citric acid, and di and tri carboxylic acids having from 21 to 54 carbon atoms;
- (b) unreacted tertiary amine having the general formula defined above, said unreacted amine being present in an amount of at least 0.2 times the stoichiometric amount of tertiary amine required to form the multi-functional carboxylic acid complex of (a); and
- (c) an alkyl cellulose ether selected from the group consisting of methylcellulose, hydroxypropyl-methyl cellulose and derivatives of hydroxyethyl cellulose wherein the terminal hydrogen of the hydroxyether group is replaced by an alkyl chain having from 10 to 24 carbon atoms, and mixtures thereof, the dispersed phase of the liquid emulsion being essentially comprised of said particles of fabric conditioning composition;
- (B) from about 0.1 to 10%, by weight of one or more emulsifying agents; and
- (C) the balance water.
2. A wash cycle or rinse cycle-additive liquid emulsion according to claim 1 wherein the tertiary amine is methyl distearyl amine.
  3. A wash cycle or rinse cycle-additive liquid emulsion according to claim 1 wherein the tertiary amine is methyl di(hydrogenated tallow) amine.
  4. A wash cycle or rinse cycle-additive liquid emulsion according to claim 1 wherein R<sub>2</sub> and R<sub>3</sub> are each an alkyl group.
  5. A wash cycle or rinse cycle-additive liquid emulsion according to claim 1 wherein the multifunctional carboxylic acid is citric acid.
  6. A wash cycle or rinse cycle-additive liquid emulsion according to claim 1 wherein said unreacted amine is present in an amount of from about 0.3 to 6 times the stoichiometric amount of tertiary amine reacted to form said complex.
  7. A wash cycle or rinse cycle-additive liquid emulsion according to claim 1 wherein the fabric conditioning composition is present in an amount of from about 5 to about 20%, by weight.
  8. A wash cycle or rinse cycle-additive liquid emulsion according to claim 1 which further contains a compound for providing anti-static properties additional to that provided by said fabric conditioning composition.
  9. A wash cycle or rinse cycle-additive liquid emulsion according to claim 1 wherein the alkyl cellulose ether is present in an amount sufficient to provide high-temperature stability to said particles of fabric conditioning composition.
  10. A wash cycle or rinse cycle-additive liquid emulsion according to claim 9 wherein the alkyl cellulose ether forms at least a partial coating upon the surface of said particles of fabric conditioning composition.
  11. A process for imparting softness and anti-static properties to fabrics comprising the step of contacting the fabrics in a laundry wash or rinse liquor with an effective amount of a particulate fabric conditioning composition having a median particle diameter greater than about 10 microns comprising:
    - (a) a fabric conditioning amount of a multi-functional carboxylic acid complex of a tertiary amine formed from the reaction of (i) a tertiary amine having the general formula:



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wherein  $R_1$  is methyl or ethyl, and  $R_2$  and  $R_3$  are each independently an aliphatic group having from 12 to 22 carbon atoms, and (ii) a multi-functional carboxylic acid selected from the group consisting of citric acid, and di and tricarboxylic acids having 21 to 54 carbon atoms;

10 (b) unreacted tertiary amine having the general formula defined above, said unreacted amine being present in an amount of at least 0.2 times the stoichiometric amount of tertiary amine required to form the multi-functional carboxylic acid complex of (a); and

15 (c) an alkyl cellulose ether selected from the group consisting of methylcellulose, hydroxypropyl-methyl cellulose and derivatives of hydroxyethyl cellulose wherein the terminal hydrogen of the hydroxyether group is replaced by an alkyl chain having from 10 to 24 carbon atoms, and mixtures thereof.

12. A process in accordance with claim 11 wherein the multifunctional carboxylic acid is citric acid.

20 13. A process in accordance with claim 11 wherein the tertiary amine is methyl di (hydrogenated tallow) amine.

14. A process in accordance with claim 11 wherein the unreacted amine is present in an amount of from about 0.3 to 6 times the said stoichiometric amount.

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15. A process for preparing a stable wash cycle or rinse cycle-additive aqueous liquid emulsion according to Claim 1 comprising the steps of:

30 (a) providing an aqueous liquid containing a mixture of (i) a fabric conditioning amount of the multi-functional carboxylic acid complex of a tertiary amine as defined in Claim 1; (ii) unreacted tertiary amine as defined in Claim 1; and (iii) an alkyl cellulose ether as defined in Claim 1;

(b) heating the aqueous liquid prior to or subsequent to step (a) to a temperature sufficient to at least melt the mixture of carboxylic acid complex and unreacted tertiary amine;

35 (c) introducing into said aqueous liquid a first portion of an emulsifying agent, said first portion being an amount selected to form upon mixing with said aqueous liquid emulsified particles containing components (i), (ii), and (iii) and having a median particle diameter greater than about 10 microns;

(d) mixing the aqueous liquid with the first portion of emulsifying agent to form an emulsion containing the aforesaid emulsified particles as the dispersed phase;

40 (e) cooling the resulting emulsion to a temperature sufficiently below the melting point of the particles formed in step (d) to at least partially solidify said particles; and

(f) introducing into the emulsion following step (e) a second portion of one or more emulsifying agents to adjust the HLB value of the emulsion to that required for high-temperature phase stability.

45 16. A process in accordance with Claim 15 wherein in step (a) the mixture of carboxylic acid complex and unreacted amine is introduced into the aqueous liquid at a temperature below its melting point and wherein subsequent to step (a) said aqueous liquid is heated to a temperature sufficiently above the melting point of the mixture.

50 17. A process in accordance with Claim 15 wherein the tertiary amine in said fabric conditioning composition is methyl di(hydrogenated) tallow amine.

18. A process in accordance with Claim 15 wherein the multifunctional carboxylic acid in said fabric conditioning composition is citric acid.

55 19. A process in accordance with Claim 15 wherein the amount of unreacted tertiary amine in the aqueous liquid is from about 0.3 to 3 times the stoichiometric amount of amine reacted to form the carboxylic acid complex.

20. A process in accordance with Claim 15 wherein the amount of unreacted tertiary amine in the liquid is

from about 5 to 30%, by weight, of said liquid.

21. A process in accordance with Claim 15 further including the step of introducing a viscosity modifier and optionally other adjuvants into said aqueous liquid.

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22. A process in accordance with Claim 15 wherein in step (d) the emulsified particles of fabric conditioning composition have a median diameter of from about 25 to 50 microns.

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## EUROPEAN SEARCH REPORT

Application Number

EP 91 20 1908

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.5)
A	EP-A-0 340 872 (COLGATE-PALMOLIVE) * the whole document * & US-A-4 869 836 * - - -	1-22	C 11 D 3/00 C 11 D 3/22 C 11 D 1/40 C 11 D 1/65
A	FR-A-2 601 390 (COLGATE-PALMOLIVE) * page 5, line 9 - page 6, line 30; claims 1-23 ** & US-A-4 828 722 * - - -	1-22	
A	EP-A-0 267 657 (PROCTER & GAMBLE) * claims 10-17 ** - - - - -	1,9-11	
			TECHNICAL FIELDS SEARCHED (Int. Cl.5)
			C 11 D
The present search report has been drawn up for all claims			
Place of search		Date of completion of search	Examiner
The Hague		28 October 91	GRITTERN A.G.
<div>CATEGORY OF CITED DOCUMENTS</div> <div><div>X : particularly relevant if taken alone</div><div>Y : particularly relevant if combined with another document of the same category</div><div>A : technological background</div><div>O : non-written disclosure</div><div>P : intermediate document</div><div>T : theory or principle underlying the invention</div></div> <div><div>E : earlier patent document, but published on, or after the filing date</div><div>D : document cited in the application</div><div>L : document cited for other reasons</div><div>&amp; : member of the same patent family, corresponding document</div></div>			